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## Section 1 - CHEMICAL PRODUCT AND COMPANY IDENTIFICATION

### PRODUCT NAME

APS BUFFER SOLUTION PH 4.0

### SYNONYMS

buffer

# PRODUCT USE

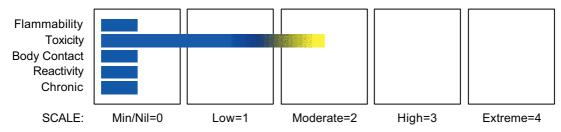
Used as a pH 4.0 buffer in pH determinations.

### Section 2 - HAZARDS IDENTIFICATION

### STATEMENT OF HAZARDOUS NATURE

NON-HAZARDOUS SUBSTANCE. NON-DANGEROUS GOODS. According to NOHSC Criteria, and ADG Code.

## CHEMWATCH HAZARD RATINGS



# POISONS SCHEDULE

None

#### RISK

- Ingestion may produce health damage\*.
- Cumulative effects may result following exposure\*.

## SAFETY

- Do not breathe gas/ fumes/ vapour/ spray.
- Avoid contact with skin.

\* (limited evidence).

Section 3 - COMPOSITION / INFORMATION ON INGREDIEN	ITS	
NAME	CAS RN	%
potassium hydrogen phthalate	877-24-7	0-10
sodium azide	26628-22-8	0-0.09
water	7732-18-5	>60
No other ingredient information supplied.		

## Section 4 - FIRST AID MEASURES

# SWALLOWED

- IF SWALLOWED, REFER FOR MEDICAL ATTENTION, WHERE POSSIBLE, WITHOUT DELAY.
- For advice, contact a Poisons Information Centre or a doctor.
- Urgent hospital treatment is likely to be needed.

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- In the mean time, qualified first-aid personnel should treat the patient following observation and employing supportive measures as indicated by the patient's condition.
- If the services of a medical officer or medical doctor are readily available, the patient should be placed in his/her care and a copy of the MSDS should be provided. Further action will be the responsibility of the medical specialist.
- If medical attention is not available on the worksite or surroundings send the patient to a hospital together with a copy of the MSDS.
- Where medical attention is not immediately available or where the patient is more than 15 minutes from a hospital or unless instructed otherwise:
- INDUCE vomiting with fingers down the back of the throat, ONLY IF CONSCIOUS. Lean patient forward or place on left side (head-down position, if possible) to maintain open airway and prevent aspiration. NOTE: Wear a protective glove when inducing vomiting by mechanical means.

# EYE

- If this product comes in contact with the eyes:
- Wash out immediately with fresh running water.
- Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids.
- · Seek medical attention without delay; if pain persists or recurs seek medical attention.
- Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.

## SKIN

- If skin or hair contact occurs:
- Flush skin and hair with running water (and soap if available).
- · Seek medical attention in event of irritation.

# INHALED

- If fumes or combustion products are inhaled remove from contaminated area.
- Other measures are usually unnecessary.

# NOTES TO PHYSICIAN

Treat symptomatically.

# Section 5 - FIRE FIGHTING MEASURES

### **EXTINGUISHING MEDIA**

■ The product contains a substantial proportion of water, therefore there are no restrictions on the type of extinguishing media which may be used. Choice of extinguishing media should take into account surrounding areas.

Though the material is non-combustible, evaporation of water from the mixture, caused by the heat of nearby fire, may produce floating layers of combustible substances.

In such an event consider:

- foam.
- dry chemical powder.
- carbon dioxide.

### **FIRE FIGHTING**

- Alert Fire Brigade and tell them location and nature of hazard.
- Wear breathing apparatus plus protective gloves for fire only.
- Prevent, by any means available, spillage from entering drains or water courses.
- Use fire fighting procedures suitable for surrounding area.
- DO NOT approach containers suspected to be hot.
- Cool fire exposed containers with water spray from a protected location.
- If safe to do so, remove containers from path of fire.
- Equipment should be thoroughly decontaminated after use.

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# FIRE/EXPLOSION HAZARD

- Non combustible.
- Not considered to be a significant fire risk.
- Expansion or decomposition on heating may lead to violent rupture of containers.
- Decomposes on heating and may produce toxic fumes of carbon monoxide (CO).
- May emit acrid smoke.

Decomposition may produce toxic fumes of: carbon dioxide (CO2), other pyrolysis products typical of burning organic material.

May emit poisonous fumes.

### FIRE INCOMPATIBILITY

None known.

HAZCHEM

None

# PERSONAL PROTECTION

Glasses: Chemical goggles. Gloves: PVC chemical resistant type.

Respirator: Type B- P Filter of sufficient capacity

## Section 6 - ACCIDENTAL RELEASE MEASURES

### **MINOR SPILLS**

- Clean up all spills immediately.
- Avoid breathing vapours and contact with skin and eyes.
- Control personal contact by using protective equipment.
- · Contain and absorb spill with sand, earth, inert material or vermiculite.
- Wipe up.
- Place in a suitable, labelled container for waste disposal.

# MAJOR SPILLS

- Moderate hazard.
- · Clear area of personnel and move upwind.
- Alert Fire Brigade and tell them location and nature of hazard.
- Wear breathing apparatus plus protective gloves.
- Prevent, by any means available, spillage from entering drains or water course.
- Stop leak if safe to do so.
- Contain spill with sand, earth or vermiculite.
- Collect recoverable product into labelled containers for recycling.
- Neutralise/decontaminate residue.
- Collect solid residues and seal in labelled drums for disposal.
- Wash area and prevent runoff into drains.
- After clean up operations, decontaminate and launder all protective clothing and equipment before storing and re-using.
- If contamination of drains or waterways occurs, advise emergency services.

# Personal Protective Equipment advice is contained in Section 8 of the MSDS.

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# Section 7 - HANDLING AND STORAGE

# PROCEDURE FOR HANDLING

- Avoid all personal contact, including inhalation.
- · Wear protective clothing when risk of exposure occurs.
- Use in a well-ventilated area.
- Prevent concentration in hollows and sumps.
- DO NOT enter confined spaces until atmosphere has been checked.
- · DO NOT allow material to contact humans, exposed food or food utensils.
- · Avoid contact with incompatible materials.
- When handling, DO NOT eat, drink or smoke.
- Keep containers securely sealed when not in use.
- · Avoid physical damage to containers.
- Always wash hands with soap and water after handling.
- Work clothes should be laundered separately. Launder contaminated clothing before re-use.
- · Use good occupational work practice.
- Observe manufacturer's storing and handling recommendations.
- · Atmosphere should be regularly checked against established exposure standards to ensure safe working conditions are maintained.

#### SUITABLE CONTAINER

- Polyethylene or polypropylene container.
- Packing as recommended by manufacturer.
- · Check all containers are clearly labelled and free from leaks.

## STORAGE INCOMPATIBILITY

None known.

# STORAGE REQUIREMENTS

- · Store in original containers.
- · Keep containers securely sealed.
- Store in a cool, dry, well-ventilated area.
- Store away from incompatible materials and foodstuff containers.
- Protect containers against physical damage and check regularly for leaks.
- Observe manufacturer's storing and handling recommendations.

# SAFE STORAGE WITH OTHER CLASSIFIED CHEMICALS



- +; May be stored together
- May be stored together with specific preventions О:
- X Must not be stored together

Section 8 - EXPOSURE CONTROLS / PERSONAL PROTECTION

**EXPOSURE CONTROLS** 

# **CHEMWATCH 4706-44** Version No:3 CD 2010/2 Page 5 of 12 Section 8 - EXPOSURE CONTROLS / PERSONAL PROTECTION

Source	Material	TWA ppm	TWA mg/m³	STEL ppm STEL mg/m³	Peak pp	m Peak mg/m³	TWA F/CC Notes
Australia Exposure Standards	sodium azide (Sodium azide (f))			· ·	0.11	0.3	
-	aterials had no OELs on Irogen phthalate:	our records		CAS:877-24-7			
• water:				CAS:7732-18-5			

#### MATERIAL DATA

POTASSIUM HYDROGEN PHTHALATE:

SODIUM AZIDE:

Sensory irritants are chemicals that produce temporary and undesirable side-effects on the eyes, nose or throat. Historically occupational exposure standards for these irritants have been based on observation of workers' responses to various airborne concentrations. Present day expectations require that nearly every individual should be protected against even minor sensory irritation and exposure standards are established using uncertainty factors or safety factors of 5 to 10 or more. On occasion animal no-observable-effectlevels (NOEL) are used to determine these limits where human results are unavailable. An additional approach, typically used by the TLV committee (USA) in determining respiratory standards for this group of chemicals, has been to assign ceiling values (TLV C) to rapidly acting irritants and to assign short-term exposure limits (TLV STELs) when the weight of evidence from irritation, bioaccumulation and other endpoints combine to warrant such a limit. In contrast the MAK Commission (Germany) uses a five-category system based on intensive odour, local irritation, and elimination half-life. However this system is being replaced to be consistent with the European Union (EU) Scientific Committee for Occupational Exposure Limits (SCOEL); this is more closely allied to that of the USA.

OSHA (USA) concluded that exposure to sensory irritants can:

cause inflammation

· cause increased susceptibility to other irritants and infectious agents

· lead to permanent injury or dysfunction

· permit greater absorption of hazardous substances and

• acclimate the worker to the irritant warning properties of these substances thus increasing the risk of overexposure.

# POTASSIUM HYDROGEN PHTHALATE:

SODIUM AZIDE:

■ It is the goal of the ACGIH (and other Agencies) to recommend TLVs (or their equivalent) for all substances for which there is evidence of health effects at airborne concentrations encountered in the workplace.

At this time no TLV has been established, even though this material may produce adverse health effects (as evidenced in animal experiments or clinical experience). Airborne concentrations must be maintained as low as is practically possible and occupational exposure must be kept to a minimum.

NOTE: The ACGIH occupational exposure standard for Particles Not Otherwise Specified (P.N.O.S) does NOT apply.

### **APS BUFFER SOLUTION PH 4.0:**

Not available

### SODIUM AZIDE:

for sodium azide:

Deaths due to sodium azide exposure are due to acute cardiovascular collapse, and central respiratory paralysis. Inhalation (or ingestion) produces dizziness, weakness, blurred vision, slight dyspnea, tachypnea, hypotension, tachycardia, acidosis, abdominal pain and spasms. Serious exposures produce polydipsia, leukocytosis, pulmonary oedema, bronchitis, convulsions, unconsciousness, and death. Solutions of sodium azide release hydrazoic acid which has the same degree of acute toxicity as the salt.

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Exposures must be controlled to very low levels to avoid hypotension.

Estimation of a maximal intake of sodium azide over an 8-hour shift (assuming 10 m3 of air inhaled and a 75% retention) for a daily pulmonary absorption of 0.75 mg results in an ambient air concentration of 0.04 ppm (0.1 mg/m3). The permissible limits of exposure to both sodium and hydrogen azides are thought to incorporate a reasonable margin of safety against headache and other symptoms of systemic discomfort and are thought to be protective against significant changes in electrocardiogram readings and hypotension. An alert has been published by OSHA to highlight the fact that significant percutaneous absorption may occur and this route may contribute significantly to worker exposure.

## WATER:

■ No exposure limits set by NOHSC or ACGIH.

# PERSONAL PROTECTION



# EYE

- · Safety glasses with side shields
- · Chemical goggles.
- Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants. A
  written policy document, describing the wearing of lens or restrictions on use, should be created for each
  workplace or task. This should include a review of lens absorption and adsorption for the class of
  chemicals in use and an account of injury experience. Medical and first-aid personnel should be trained in
  their removal and suitable equipment should be readily available. In the event of chemical exposure, begin
  eye irrigation immediately and remove contact lens as soon as practicable. Lens should be removed at the
  first signs of eye redness or irritation lens should be removed in a clean environment only after workers
  have washed hands thoroughly. [CDC NIOSH Current Intelligence Bulletin 59].

# HANDS/FEET

- Wear chemical protective gloves, eg. PVC.
- Wear safety footwear or safety gumboots, eg. Rubber.

# OTHER

- Overalls.
- P.V.C. apron.
- Barrier cream.
- Skin cleansing cream.
- Eye wash unit.

## RESPIRATOR

■ Selection of the Class and Type of respirator will depend upon the level of breathing zone contaminant and the chemical nature of the contaminant. Protection Factors (defined as the ratio of contaminant outside and inside the mask) may also be important.

Breathing Zone Level	Maximum Protection	Half- face Respirator	Full- Face Respirator
ppm (volume)	Factor		
1000	10	B- AUS P	-
1000	50	-	B- AUS P
5000	50	Airline *	-

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5000	100	_	B- 2 P
10000	100	-	B- 3 P
	100+		Airline**

\* - Continuous Flow \*\* - Continuous-flow or positive pressure demand.

The local concentration of material, quantity and conditions of use determine the type of personal protective equipment required. For further information consult site specific CHEMWATCH data (if available), or your Occupational Health and Safety Advisor.

### **ENGINEERING CONTROLS**

NC317TCP

■ General exhaust is adequate under normal operating conditions. If risk of overexposure exists, wear SAA approved respirator. Correct fit is essential to obtain adequate protection. Provide adequate ventilation in warehouse or closed storage areas. Air contaminants generated in the workplace possess varying "escape" velocities which, in turn, determine the "capture velocities" of fresh circulating air required to effectively remove the contaminant.

Type of Contaminant: solvent, vapours, degreasing etc., evaporating from tank (in still air)	Air Speed: 0.25- 0.5 m/s (50- 100 f/min)
aerosols, fumes from pouring operations, intermittent container filling, low speed conveyer transfers, welding, spray drift, plating acid fumes, pickling (released at low velocity into zone of active generation)	0.5- 1 m/s (100- 200 f/min.)
direct spray, spray painting in shallow booths, drum filling, conveyer loading, crusher dusts, gas discharge (active generation into zone of rapid air motion)	1- 2.5 m/s (200- 500 f/min)
grinding, abrasive blasting, tumbling, high speed wheel generated dusts (released at high initial velocity into zone of very high rapid air motion).	2.5- 10 m/s (500- 2000 f/min.)

Within each range the appropriate value depends on:

Lower end of the range 1: Room air currents minimal or favourable to	Upper end of the range 1: Disturbing room air currents
capture 2: Contaminants of low toxicity or of nuisance value only	2: Contaminants of high toxicity
<ul><li>3: Intermittent, low production.</li><li>4: Large hood or large air mass in motion</li></ul>	3: High production, heavy use 4: Small hood - local control only

Simple theory shows that air velocity falls rapidly with distance away from the opening of a simple extraction pipe. Velocity generally decreases with the square of distance from the extraction point (in simple cases). Therefore the air speed at the extraction point should be adjusted, accordingly, after reference to distance from the contaminating source. The air velocity at the extraction fan, for example, should be a minimum of 1-2 m/s (200-400 f/min.) for extraction of solvents generated in a tank 2 meters distant from the extraction point. Other mechanical considerations, producing performance deficits within the extraction apparatus, make it essential that theoretical air velocities are multiplied by factors of 10 or

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more when extraction systems are installed or used.

### Section 9 - PHYSICAL AND CHEMICAL PROPERTIES

#### APPEARANCE

Colourless solution, red colour coded solution; mixes with water.

#### PHYSICAL PROPERTIES

Liquid. Mixes with water.

State	Liquid	Molecular Weight	Not applicable
Melting Range (°C)	Not available	Viscosity	Not Available
Boiling Range (°C)	Not available	Solubility in water (g/L)	Miscible
Flash Point (°C)	Not applicable	pH (1% solution)	Not available
Decomposition Temp (°C)	Not available	pH (as supplied)	4.0
Autoignition Temp (°C)	Not applicable	Vapour Pressure (kPa)	Not available
Upper Explosive Limit (%)	Not applicable	Specific Gravity (water=1)	1.0
Lower Explosive Limit (%)	Not applicable	Relative Vapour Density (air=1)	Not available
Volatile Component (%vol)	Not available	Evaporation Rate	Not available
	NUL avallable	Evaporation Rate	inor available

# Section 10 - CHEMICAL STABILITY

# CONDITIONS CONTRIBUTING TO INSTABILITY

- Presence of incompatible materials.
- Product is considered stable.
- Hazardous polymerisation will not occur.

For incompatible materials - refer to Section 7 - Handling and Storage.

### Section 11 - TOXICOLOGICAL INFORMATION

# POTENTIAL HEALTH EFFECTS

### ACUTE HEALTH EFFECTS

### SWALLOWED

Accidental ingestion of the material may be harmful; animal experiments indicate that ingestion of less than 150 gram may be fatal or may produce serious damage to the health of the individual. Azides, if swallowed, can cause breathlessness and rapid heart beat within 5 minutes. Nausea, vomiting, headache, restlessness and diarrhoea can occur within 15 minutes. Other symptoms include low blood pressure which cannot be corrected, abnormal breathing, reduced body temperature, reduced blood pH, convulsions, collapse and death. Continued administration can cause increased sensitivity. Poisoning can cause headaches and acidosis. Several grams of sodium azide can cause liver, pulmonary and brain swelling with death occurring in less than an hour. Large doses of azide increases the blood pressure and causes generalised convulsions, followed by depression and collapse.

### EYE

■ Although the liquid is not thought to be an irritant (as classified by EC Directives), direct contact with the eye may produce transient discomfort characterised by tearing or conjunctival redness (as with windburn).

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# SKIN

Skin contact is not thought to produce harmful health effects (as classified under EC Directives using animal models). Systemic harm, however, has been identified following exposure of animals by at least one other route and the material may still produce health damage following entry through wounds, lesions or abrasions. Good hygiene practice requires that exposure be kept to a minimum and that suitable gloves be used in an occupational setting.

Entry into the blood-stream, through, for example, cuts, abrasions or lesions, may produce systemic injury with harmful effects. Examine the skin prior to the use of the material and ensure that any external damage is suitably protected.

## INHALED

■ Not normally a hazard due to non-volatile nature of product.

The material is not thought to produce either adverse health effects or irritation of the respiratory tract following inhalation (as classified by EC Directives using animal models). Nevertheless, adverse systemic effects have been produced following exposure of animals by at least one other route and good hygiene practice requires that exposure be kept to a minimum and that suitable control measures be used in an occupational setting.

# CHRONIC HEALTH EFFECTS

■ Long-term exposure to the product is not thought to produce chronic effects adverse to the health (as classified by EC Directives using animal models); nevertheless exposure by all routes should be minimised as a matter of course.

# TOXICITY AND IRRITATION

SODIUM AZIDE:

POTASSIUM HYDROGEN PHTHALATE:

■ unless otherwise specified data extracted from RTECS - Register of Toxic Effects of Chemical Substances.

■ Not available. Refer to individual constituents.

POTASSIUM HYDROGEN PHTHALATE:

# TOXICITY

# IRRITATION

Oral (Rat) LD50: >3200 mg/kg Dermal (Guinea pig) LD50: >1000 mg/kg

■ Asthma-like symptoms may continue for months or even years after exposure to the material ceases. This may be due to a non-allergenic condition known as reactive airways dysfunction syndrome (RADS) which can occur following exposure to high levels of highly irritating compound. Key criteria for the diagnosis of RADS include the absence of preceding respiratory disease, in a non-atopic individual, with abrupt onset of persistent asthma-like symptoms within minutes to hours of a documented exposure to the irritant. A reversible airflow pattern, on spirometry, with the presence of moderate to severe bronchial hyperreactivity on methacholine challenge testing and the lack of minimal lymphocytic inflammation, without eosinophilia, have also been included in the criteria for diagnosis of RADS. RADS (or asthma) following an irritating inhalation is an infrequent disorder with rates related to the concentration of and duration of exposure to the irritating substance. Industrial bronchitis, on the other hand, is a disorder that occurs as result of exposure due to high concentrations of irritating substance (often particulate in nature) and is completely reversible after exposure ceases. The disorder is characterised by dyspnea, cough and mucus production.

# SODIUM AZIDE:

TOXICITY Oral (man) LDLo: 143 mg/kg Oral (human) TDLo: 0.71 mg/kg Oral (woman) TDLo: 3 mg/kg IRRITATION Nil Reported

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Oral (rat) LD50: 27 mg/kg Dermal (rabbit) LD50: 20 mg/kg General anaesthesia, somnolence, convulsions, headache, irritability, arrhythmias, dyspnae, respiratory stimulation, diarrhoea recorded.

WATER:

• No significant acute toxicological data identified in literature search.

### Section 12 - ECOLOGICAL INFORMATION

Refer to data for ingredients, which follows:

POTASSIUM HYDROGEN PHTHALATE: SODIUM AZIDE: APS BUFFER SOLUTION PH 4.0: DO NOT discharge into sewer or waterways.

SODIUM AZIDE:

■ Very toxic to aquatic organisms, may cause long-term adverse effects in the aquatic environment.

Do NOT allow product to come in contact with surface waters or to intertidal areas below the mean high

water mark. Do not contaminate water when cleaning equipment or disposing of equipment wash-waters.

Wastes resulting from use of the product must be disposed of on site or at approved waste sites.

Toxicity Fish: 24hr LC50: 5000 ppm

Toxicity Fish (L. macrochirus) 24hr LC50: 1.5 ppm \* Acute toxicity for lower organisms: 5 mg/l approx. \* Acute toxicity for cold-blooded animals: 1 mg/l \* Microorganisms LC50: 5 mg/l \*

\* [Merck]

Ecotoxicity				
Ingredient	Persistence:	Persistence: Air	Bioaccumulation	Mobility
-	Water/Soil			-
sodium azide	LOW		LOW	HIGH
water	LOW		LOW	HIGH

# Section 13 - DISPOSAL CONSIDERATIONS

• Recycle wherever possible.

- Consult manufacturer for recycling options or consult local or regional waste management authority for disposal if no suitable treatment or disposal facility can be identified.
- Dispose of by: burial in a land-fill specifically licenced to accept chemical and / or pharmaceutical wastes or incineration in a licenced apparatus (after admixture with suitable combustible material).
- Decontaminate empty containers. Observe all label safeguards until containers are cleaned and destroyed.
- Containers may still present a chemical hazard/ danger when empty.
- Return to supplier for reuse/ recycling if possible.

Otherwise:

- If container can not be cleaned sufficiently well to ensure that residuals do not remain or if the container cannot be used to store the same product, then puncture containers, to prevent re-use, and bury at an authorised landfill.
- Where possible retain label warnings and MSDS and observe all notices pertaining to the product.

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## Section 14 - TRANSPORTATION INFORMATION

HAZCHEM: None (ADG7)

NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS: UN, IATA, IMDG

Section 15 - REGULATORY INFORMATION

# POISONS SCHEDULE

None

**REGULATIONS** Regulations for ingredients

## potassium hydrogen phthalate (CAS: 877-24-7) is found on the following regulatory lists;

"Australia Inventory of Chemical Substances (AICS)"

### sodium azide (CAS: 26628-22-8) is found on the following regulatory lists;

"Australia Exposure Standards", "Australia Hazardous Substances", "Australia Inventory of Chemical Substances (AICS)", "OECD Representative List of High Production Volume (HPV) Chemicals"

### water (CAS: 7732-18-5) is found on the following regulatory lists;

"Australia Inventory of Chemical Substances (AICS)","IMO IBC Code Chapter 18: List of products to which the Code does not apply","OECD Representative List of High Production Volume (HPV) Chemicals"

# No data for APS Buffer Solution pH 4.0 (CW: 4706-44)

Section 16 - OTHER INFORMATION

# Denmark Advisory list for selfclassification of dangerous substances

Substance	CAS	Suggested codes
potassium hydrogen phthalate	877-24-7	Xn Mut3; R68
		Repr3; R63

### EXPOSURE STANDARD FOR MIXTURES

■ "Worst Case" computer-aided prediction of spray/ mist or fume/ dust components and concentration:

■ Composite Exposure Standard for Mixture (TWA) :100 mg/m<sup>3</sup>.

■ Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references.

A list of reference resources used to assist the committee may be found at: www.chemwatch.net/references.

■ The (M)SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering

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Hazard Alert Code: MODERATE

controls must be considered.

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This is the end of the MSDS.